TOTAL SYNTHESIS OF 6-OXO-GRINDELIC ACID METHYL ESTER^{†)}

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The first synthesis of the methyl ester of 6-oxo-grindelic acid (1) from decalone derivative 2 through γ -hydroxyenone 7 was described.

6-Oxo-grindelic acid $(\underline{1})$ isolated from <u>Grindelica robsta¹⁾</u> is a highly oxidized labdane type diterpene. The presence of the adjacent oxidized functional groups of this molecule aroused interest of its synthesis.

Recently we have developed²⁾ the two new synthetic methods of 9-carbomethoxy- $\Delta^{6\,(7)}$ -4,4,10-trimethyl decalone (2), one of which is the direct photocyclization of 10-carbomethoxy- β -ionone enolate and the other is the thermal electrocyclic reaction of 7,8-cis-10-carbomethoxy- β -ionone enolacetate followed by methanolysis. Compound 2 bears exactly adequate functional groups accessible to the synthesis of 6-oxo-grindelic acid (1). Thus 6-oxo function could be introduced by the oxidative rearrangement of chromic ester³⁾ of t-8- β -hydroxyl group which is generated by methylation of the carbonyl at C-8, and SeO₂ oxidation of the resulted enone could yield the tertiary hydroxyl function required for the ether linkage at C-9.

Compound $\underline{2}$ was converted to exomethylene enone $\underline{3}^{4}$) (LiAlH₄, MnO₂, H⁺, 88 % yield) which was derived to diketone 4 (methyl acetoacetate/catalytic NaOMe/MeOH, 1.3 equiv. of 1 % NaOH/H₂O-MeOH, H⁺-heating, 81 % overall yield). $8-\beta$ -Hydroxyl derivative $\underline{5}$ was obtained by reduction (LiAlH₄) of $\underline{4}$ followed by selective oxidation (MnO₂), methylation (excess MeLi, ether)⁵⁾, and acetylation (Ac₂O-Py) (80 % yield from $\underline{4}$). The successful oxidative rearrangement of $\underline{5}$ to enone $\underline{6}$ was achieved by CrO_3-5 % aq H_2SO_4 in ether (0°C, 2 h, 84 % yield) and then SeO_2 oxidation of $\underline{6}$ afforded γ -hydroxyenone $\underline{7}$ in 41 % yield (dioxane, reflux, 30 m) $\overline{}^{5)}$. C-9-hydroxyl group was protected as tetrahydropyranyl ether (dihydropyran, PPTS) prior to hydrolysis (aq K_2CO_3 , MeOH) of acetoxyl group and following oxidation ($CrO_3 \cdot 2Py$) to avoid the formation of hemiacetal 8 which did not react with Wittig-Horner reagent. 9 (48 % from 7) afforded unsaturated ester 10 by Wittig-Horner reaction ((CH₃O)₂POCH₂CO₂CH₃, NaH, THF, quantitative yield) followed by removing the protecting group (70 % AcOH, room temp., overnight). The Z isomer of $\underline{10}$ was less than 5 % by the nmr spectrum. Finally the synthesis of 6-oxo-grindelic acid methyl ester (11), mp 71-73°C, was achieved by a treatment of 10 with triethylamine (reflux, 24 h, ~ 50 % yield for the last two steps) 6). The nmr and ir spectra of synthesized 11 were identical with those reported for the methyl ester of natural compound 1.

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References

- †) Dedicated to Emeritus Professor Takeo Sakan on the occasion of his 70th birthday.
- 1) L. Mangoni and M. Balardini, Gazz. Chim. Ital., 92, 983 (1962).
- 2) The details will be reported elsewhere.
- 3) W. G. Dauben and D. M. Michno, J. Org. Chem., 42, 682 (1977).
- 4) Satisfactory spectral data were obtained for all new compounds.
- 5) Both methyl at C-8 and hydroxyl at C-9 must be introduced exclusively from the less hindered α -side.
- 6) This reaction afforded the one stereoisomer predominantly as a major product. The stereochemistry at the C-13 position of the methyl ester of natural compound (1) is unknown.